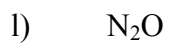
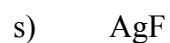
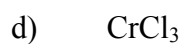
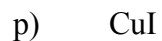


Chapter 2: Worksheet #2 Nomenclature

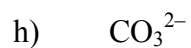
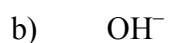
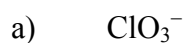
1. Name the following binary compounds:



2. Name the following acids:



3. Name these polyatomic ions:



4. Name the following compounds that contain polyatomic ions:



5. Write the formula for the chemical name below:

a) lithium bromide

g) phosphorus triiodide

b) diboron trioxide

h) tin(IV) sulfide

c) mercury(II) sulfate

i) barium sulfate

d) nitrous acid

j) hydrobromic acid

e) aluminum hydroxide

k) sodium hydrogen sulfite

f) perbromic acid

l) potassium sulfide

6. Write the formula for the oxide below:

a) iron(II)

c) aluminum

b) iron(III)

d) magnesium